ISSN:1998-0507 doi: https://ejesm.org/doi/v18i3.1 Submitted: January 28, 2025 Accepted: July 18, 2025

PHOSPHATE SPECIATION IN SEDIMENTS OF CAPE TOWN WATERWAYS- AN INSIGHT INTO ANTHROPOGENIC INPUTS

OPUTU, O.U.¹ AND *AKHARAME, M.O.²

¹The Hair and Skin Research Laboratory, Division of Dermatology, Department of Medicine, University of Cape Town, Cape Town, 8000, South Africa ²Department of Environmental Management and Toxicology, University of Benin, Benin-City, 3002, Nigeria

*Corresponding author: michael.akharame@uniben.edu

Abstract

This study investigated the levels and distribution of phosphorus in sediments along the Swart River, Cape Town, and identified the pollution trends along the river. The harmonised protocol developed by the Standards Measurements and Testing Program of the European Commission was used to extract the different fractions of phosphorus in the sediments. The phosphorus levels in the extracts were measured with a spectrophotometer operated at 880 nm. The total phosphorus levels ranged from 797.7 – 6877.1 ug/g with the inorganic phosphorus being the principal constituent of the total extractable phosphorus in most of the sampling sites surveyed. The speciation results showed that HCl-P was the dominant form of inorganic phosphorus, while NaOH-P was the minor non-apatitic constituent in the sediments. Most of the inorganic phosphorus was fixed in the apatitic form. The total phosphorus levels in the sediments are indicative of high pollution along the Swart River, with the deposition majorly from anthropogenic sources (industrial activities, sewage/effluent discharges, and agricultural run-off). Eutrophication problems could arise at some of the sampling points in the future if the prevailing conditions are not mitigated.

Keywords: Phosphorus, Sediments, Speciation, Swart River, Cape Town

Introduction

Phosphorus is an essential macronutrient required by all living organisms due to its involvement in many biochemical processes ((Bains *et al.*, 2019). It is considered a limiting nutrient for primary production in terrestrial and aquatic ecosystems (Elser, 2012). The occurrence of excess phosphorus in fresh surface waters is undesirable as it accelerates eutrophication (Prüter *et al.*,

2020; Heathwaite and Sharpley, 1999). This phenomenon dramatically reduces the quality of water and adversely impacts aquatic ecosystems. Thus, the leaching of nutrients such as phosphorus from anthropogenic and natural sources must be carefully monitored and controlled where possible (Li *et al.*, 2020; Paytan and McLaughlin, 2007).

Phosphorus in surface water bodies results from several anthropogenic

activities such as domestic waste dumping, industrial discharges, agricultural waste discharges, and storm run-offs (McDonald et al., 2019; Li et al., 2018; Owa, 2014). Phosphorus speciation in water systems occurs in various chemical forms, establishing a dynamic equilibrium between the dissolved and particulate state $(P_{dissolved} \leftrightarrow P_{particulate})$ (Spivakov et al., 1999). The sediment of water bodies acts as a sink for the entrapment of phosphorus (Acharya et al., 2016) and usually consists of embedded organic and inorganic forms (Li et al., 2020). However, the entrapped phosphorus in sediments may further fuel the eutrophication process or increase phosphorus levels in water bodies as changes in the water chemistry can mobilise the sequestered phosphorus ((Bastami et al., 2018; Mainstone and Parr, 2002; Bradford and Peters, 1987). The phosphorus enrichment is favoured under anaerobic conditions substantial pH increases in water systems ((Mao et al., 2021; Spivakov et al., 1999). Generally, agitations in water bodies leading to the re-suspension of sediments and chemical conditions such as pH, dissolved oxygen, surface charges, and variation in iron concentrations can accelerate the deposition of phosphorus into the overlying water column (Liu et al., 2016; Zhou et al., 2016; Meng et al., 2014; Coelho et al., 2004).

Phosphorus mainly occurs in forms such as the loosely-sorbed-P, Fe-bound-P, authigenic/biogenic-P, detrital apatite-P and organic-P (Bastami *et al.*, 2018; Ruttenberg, 1992). The nature of these species (i.e. biological and chemical activity) determines whether the phosphorus is reactive or refractory. This has crucial implications for mobility and

bioavailability (Dan et al., 2020; Bastami et al., 2018). The levels of the different fractionated phosphorus species in water sediments play a critical role in the overall health of water bodies (Yang et al., 2019; Li et al., 2018). Due to the relationship between sediment phosphorus concentration and surface water quality, it has become pertinent to monitor various phosphorus species, especially in water bodies where such information is scarce.

This study represents a general perspective of speciated phosphorus forms in sediments of the Swart River in Cape Town. South Africa. Major anthropogenic points and contributors along the course of the river were identified to evaluate phosphorus distribution and thus the contribution of human activities and other sources that may increase the phosphorus in river modified sediments. The William's protocol adopted by the Standards, Measurements and Testing Programme BCR) of the European (formerly Commission to extract phosphorus was utilised in this investigation (Ruban et al., 1999). The protocol enabled the extraction of total phosphorus (TP), organic phosphorus (OP), inorganic phosphorus (IP), NaOH phosphorus - NaOH-P (P bound to Al. Fe and Mn oxides and hydroxides) and HCl phosphorus – HCl-P (P associated with Ca) from the river sediments (Ruban et al., 2001). The validation of the extraction process was done by using certified reference material (BCR-684). The results obtained for the ten replicates showed no significant difference (p<0.005) from the values in the certificate of analysis.

Study Area

The City of Cape Town is the provincial capital of Western Cape in

South Africa, and also the legislative capital of the country (34° 0'18° 30'E). The Atlantic Ocean flanks the western and southern parts of the city, while the eastern region is flanked by the Indian Ocean (Oputu and Akharame, 2022). One of the rivers that flows through the city and empties into the Atlantic Ocean is the Swart River. The Swart River receives water from Langa (source Epping) and Bridgetown (source Heideveld) passes through the Athlone treatment plant, Mowbray golf course,

converges with the Liesbeek River and continues its path towards the Atlantic Ocean. Along the route of the river, it is exposed to various forms of pollution such as the indiscriminate dumping of refuse from informal settlements (Langa), industrial waste (Bridgetown), fertiliser run-off (Mowbray golf course), as well as, effluent discharges from Athlone wastewater treatment plant. The location of the sampling points is represented in Figure 1.

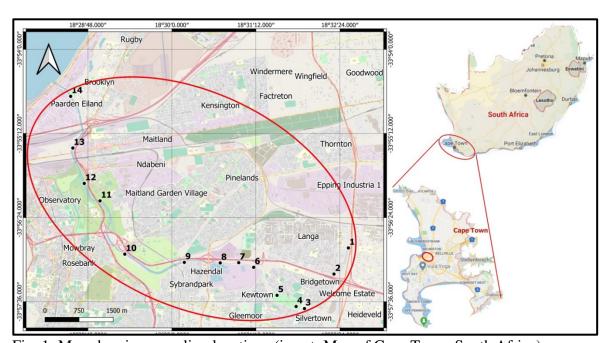


Fig. 1: Map showing sampling locations (insert: Map of Cape Town, South Africa)

Sample Collection and Extraction

A description of the 14 sampling points in the Swart River is presented in Table 1. The river sediment was collected from each of the sampling points with a PVC grab sampler (10 cm diameter x 5 cm). The different sediment scoops from each location were pooled together to cater to the within-site variability. This was followed by storing them in pre-

labelled clean zip-lock bags. Thereafter, the sediments were transported to the laboratory and air-dried at room temperature, and dried sediments were sieved (0.15 mm sieve). Phosphate-free soap was utilised for the washing of all glassware used in the analysis before rinsing with copious amounts of distilled water (Onianwa *et al.*, 2013).

Table 1: Site description

Sampling point	Location number	Site activity (human)	Site activity (industry)	Sediment type
Lange	1	High	Low	Sandy
Langa	2	High	Low	Sandy
	3	Low	High	Clay
Duidentaryn	4	Low	High	Clay
Bridgetown	5	Low	Low	Sandy
	6	Low	High	Sandy
	7	Low	High	Sandy/clay
Athlone WWTP and	8	Moderate	High	Sandy
conjunctions	9	Moderate	High	Sandy/clay
	10	High	Low	Sandy
Marshway cale agrees	11 High	High	Low	Sandy/clay
Mowbray golf course	12	High	Low	Clay
Along Swart River toward the	13	Low	Moderate	Sandy
ocean	14	Low	Moderate	Sandy

The extraction of the different fractions of phosphorus in the sediments was done using the harmonised protocol developed by the Standards Measurements and Testing Program of the European Commission, SMT protocol (Ruban et al., 2001). The fractions of phosphorus extracted using the protocol include TP, OP, IP, NaOH-P, and HCl-P. The extraction process involves sediment contacting the with appropriate extractant, shaking for a specified period, and separating the extractant and sediment by centrifuging the mixture (Figure 2). Briefly, the sediment (0.2 g) was calcined at 450°C for three hours for TP extraction. The residue was then mixed with 20 mL of 3.5 M HCl and shaken for 16 h. TP was obtained from the extract. For organic and inorganic extraction of phosphorus, the sediment

(0.2 g) was mixed with 20 mL of 1 M HCl and shaken for 16 h and IP was obtained from the extract. For OP, the residue obtained from the extract was washed (saturated NaCl solution), dried (80°C), and then calcined (450°C for 3 h). After that, the residue was mixed with 20 mL of 1 M HCl and shaken for 16 h and the OP fraction was obtained from the extract. For NaOH-P, the sediment (0.2 g) was mixed with 20 mL of 1 M NaOH and shaken for 16 h. The extract was mixed with 3.5 M HCl (4 mL) and then left to stand for 16 h. The NaOH-P was obtained from the extract. For HCl-P, the sediment (0.2 g) was mixed with 20 mL of 1 M NaOH and shaken for 16 h. The residue obtained from the extract was washed with NaCl, then mixed with 1 M HCl (20 mL) and shaken for 16 h. HCl-phosphate was obtained from the extract.

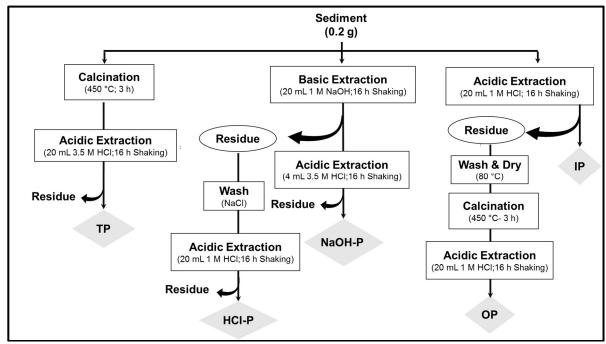


Fig. 2: Various extraction methods of phosphorus species

Phosphorus Analysis

To analyse the phosphorus content of the sediments, the Murphy and Riley Molybdenum Blue reagent was employed (Kovar and Pierzynski, 2009; Ruban et al., 2001; Murphy and Riley, 1962). 2 mL aliquots of supernatants were transferred into a 50 mL volumetric flask, and the pH of all samples adjusted to neutral, using 5 M NaOH. P-nitrophenol was used as an indicator (colour change from yellow to colourless), and 10 mL of the mixed reagent was added to develop a blue colour. Each solution was mixed, and the volume was brought to the mark with distilled water. The colour developed within 10 min, and the absorbance was measured immediately using a Cary UVspectrophotometer Vis (Agilent Technologies, Santa Clara, USA) operated at 880 nm.

Quality Control and Quality Assurance

The validity of the P extraction and analysis method was established by

assaying the certified reference material – BCR-684 purchased from Sigma Aldrich (South Africa). The extraction method and the results obtained from the analysis were compared statistically. A t-test was done on the results at a 95% confidential level to ascertain the significant difference between the certified results and the results obtained from the analysis.

Results and Discussion Quality Control and Quality Assurance Result

The results obtained from the analysis and the BCR-certified results are presented in Table 2. The t-test report showed that all the t-critical values were less than the values recorded for t-stat (Table 3). Therefore, there is no significant difference between the results from the analysis and the certified results. The methods used for analysis were validated and proven to be reproducible.

Table 2: Table showing experimental results versus BCR-analysis

	NaOH-P	in	HCl-P in		Inorganic	-P in	Organic-l	P in	Total-P in	<u>l</u>
	ug/g		ug/g		ug/g		ug/g		ug/g	
Replicate	Analysis	BCR	Analysis	BCR	Analysis	BCR	Analysis	BCR	Analysis	BCR
1	530	576	593	486	1201	1110	168	201	1574	1339
2	625	527	557	569	1001	1126	178	211	1295	1391
3	670	527	575	457	1125	1044	216	190	1522	1423
4	500	568	443	527	1959	1072	211	221	1464	1264
5	482	489	459	528	1059	1161	222	188	1334	1386
6	562	544	429	445	1324	1111	213	193	1398	1314
7	508	582	525	562	1158	1095	226	193	1206	1268
8	549	517	451	515	1025	1125	193	214	1130	1407
9	521	587	601	573	1120	1130	203	221	1295	1382
10	493	588	435	545	1158	1146	236	229	1273	1427

Table 3: The t-test results at 95% confidential level.

	T-		
P-fractionation	critical	T-stat	Comment
NaOH-P	1.761	0.3776	No significant difference between the two means
HCl-P	1.724	0.057	No significant difference between the two means
Inorganic-P	1.761	0.1507	No significant difference between the two means
Organic-P	1.73	0.4554	No significant difference between the two means
Total-P	1.717	0.1116	No significant difference between the two means

Total Phosphorus

Table 4 shows the concentration of TP, OP. IP. NaOH-P. and HCl-P. Total phosphate (TP) varied significantly in all the samples, and this could be attributed to the different sampling points taken along the Swart River. The TP levels in the samples were 1014 - 1186 ug/g (Langa), 797 - 3833 ug/g (Bridgetown), 1645 -3468 ug/g (Athlone water treatment plant), 2308 - 2475 ug/g (Mowbray golf course), and 1034 - 6877 ug/g (towards the ocean). The TP concentrations obtained from the various sites were primarily influenced by external factors (McDonald et al., 2019; Li et al., 2018). In Langa, an informal township settlement with inadequate sanitary/toiletry (Philander, 2015), the waterway is used as a dumping site for domestic waste. As such, the anthropogenic activity could account for the high TP. Bridgetown and the Athlone water treatment plant had TP in the range of 797-3833 ug/g and 16453468 ug/g, respectively. Both sampling sites are located in industrial areas with minimal human activity, and their waterway is characterised by clayey sediment. Industrial seepage and clayey sediment contribute immensely to TP as the latter absorbs more phosphate than sandy sediments.

Mowbray golf course had TP in the range of 2308 - 2475 ug/g, and the presence of TP may be hugely attributed to the fertiliser residues that were washed off the golf course (Thin et al., 2020). Fertilisers are composed of nitrogen, phosphorus. potassium and phosphorus component contributed immensely to the TP in the samples. Furthermore, the sediment in this sampling point was clayey and hence could retain more phosphorus. Along the Swart River, towards the ocean, the TP ranged from 1034 to 6877 ug/g. This sampling site contained the highest concentration of TP. This could be because the sampling point was at the meeting of the Swart River (Langa with high anthropogenic activity) and the Liesbeek River (high industrial activity). The high TP observed was possibly a combination of the phosphate from both rivers.

Generally, anthropogenic activities contribute to the elevation of phosphorus in sediments and the overlying waters. Typically, anthropogenic phosphorus is deposited in the aquatic environment via industrial activities (phosphoric acid/fertiliser production, metal production/plating), sewage/effluent discharges, agricultural run-off (farms and lawns/golf courses) and run-offs from animal farms (dairy and piggery) (DFFE, 2022). The sampling locations in this study have one of the above anthropogenic activities taking place which may be responsible for the elevated levels of phosphorus recorded in the sediments.

Table 4: Concentrations in ug/g for total-P, inorganic-P, organic-P, NaOH-P, and HCl-P

Site	Total-P	Inorganic-P	Organic-P	NaOH-P	HCl-P			
LANGA					_			
1	1014.7±13.14 ^b	553.0±27.52°	366.0 ± 26.83^{a}	113.5±15.59a	1074.0±53.76 ^{de}			
2	1186.2±21.82 ^c	418.0±31.99 ^b	526.0±36.67°	222.9±33.76 ^b	679.5±37.16 ^b			
BRIDGE	TOWN							
3	3833.8±35.09 ^j	2583.8±61.76i	514.8±24.63bc	351.5±35.75 ^{cd}	1924.3±66.43g			
4	3795.0±39.01 ^j	3187.0±65.42k	535.2±41.92°	212.3±28.56b	574.9±28.27a			
5	797.2±26.48a	105.1±11.17 ^a	680.9 ± 46.70^{d}	506.0±42.00e	2233.3±92.53i			
ATHLO	ATHLONE WATER TREATMENT PLANT AND CONJUNCTION							
6	1882.7±43.57e	1222.8±36.78 ^f	612.4±22.17 ^d	289.1±27.41c	800.9±39.70°			
7	3243.4±48.25 ^h	2479.4±66.90 ^h	734.1±47.94 ^e	378.5±50.26 ^d	2097.7±77.94 ^h			
8	3468.2±38.29i	2815.8±34.49 ^j	320.4±31.26a	609.0±38.04 ^f	1320.4±39.94 ^f			
9	1645.7±28.09 ^d	801.7 ± 28.82^{d}	507.9±37.47bc	112.6±13.16 ^a	2128.0±38.98h			
MOWBRAY GOLF COURSE								
10	2308.6±40.80 ^f	843.0±33.14 ^d	1413.6±47.94 ^f	382.9±41.73 ^d	543.2±32.30 ^a			
11	2475.2±60.99g	1915.1±36.99g	513.7±40.43bc	353.2±36.78 ^{cd}	986.7±55.26 ^d			
ALONG SWART RIVER TOWARD THE OCEAN								
12	6877.1±70.37 ^k	3420.7 ± 44.93^{1}	3418.7±88.16g	1012.4±73.79g	2390.0±73.54 ^j			
13	1034.8±23.34 ^b	404.9±22.32 ^b	534.8±49.10°	350.0±18.50 ^{cd}	1161.2±37.31e			
14	1579.6±27.68d	1131.5±39.95e	442.1±37.23 ^b	307.7±34.92°	801.0±24.26°			

Values are means \pm deviation, Different superscripts in the same column indicate Significant differences at p < 0.05 according to the Duncan Multiple Range Test (DMRT).

The phosphorus levels recorded in this study align with those reported in similar studies by various scholars. A comparison of the phosphorus speciation levels recorded in the literature for some rivers, lakes, and coastal sediments and the present study is presented in Table 5. Phosphorus levels reported for oceanic continental sediments range from 248 to 3345 µg P/g (Filippelli, 1997). The Chinese Environmental Dredging Standards stipulate that TP >500 µg P/g in sediment is considered heavily polluted

(Liu *et al.*, 1999). Also, the province of Ontario in Canada benchmarked a phosphorus level of 600 μ g P/g for marginally clean or unpolluted sediments, while contaminated sediments are > 600 μ g/g (Persaud *et al.*, 1993. Currently, there is no background level set by the South African government for assessing phosphorus contamination in sediments. As such, by using the Chinese and Canadian guidelines, it can be inferred that the Swart River is heavily polluted.

Table 5: Levels of phosphorus speciation found in the literature for some river/coastal

sediments and the present study

Study area	Total-P	Inorganic-P	References
	(µg P/g)	(μg P/g)	
Seto Island Sea, Japan	525 – 1005		Yamada and Kayama (1987)
St. Lawrence estuary, Gulf of St. Laurent	1627		Sundby et al. (1992)
Aarhus Bay, Denmark	930 - 1550		Jensen and Thamdrup (1993)
Onondaga Lake, New York, USA	2500 - 3090		Penn and Auer (1997)
Chang Jiang River (Yangtze River), China	170 - 705	135 - 619	Rao and Berner (1997)
Southern and Eastern North Sea	93 - 806		Slomp <i>et al.</i> (1998)
Huanghe River, China	600	486	Yue and Ji-Jin (1999)
Huanghe Estuary, China	806	538	
Huanghe Shelf, China	580	512	
Bohai sea, China	322 - 617	252 - 524	
Yellow Sea, China	233 - 512	174 - 481	Liu et al. (2004)
Ratones River, Southern Brazil	4650 - 11780		Pagliosa et al. (2005)
Tavares River, Souther Brazil	9300 - 14880		
Verissimo River, Southern Brazil	3100 - 6200		
Aririu River, Southern Brazil	20150 - 20460		
Itacorubi River, Southern Brazil	4340 - 21080		
Maruim River, Southern Brazil	2945 - 20150		
Yellow River, China		1187 - 1468	Li and Guo (2006)
Louros River, NW Greece	240 - 620	80 - 380	Katsaounos et al. (2007)
Burclar Bay, South Anatolia, Turkey	210 - 206	202 - 205	Aydin et al. (2009)
Yangtze River, China	452 - 1140	355 – 956	Hou et al. (2009)
NE Mediterranean Sea, Antalya, Turkey	152 - 275	150 - 261	Gunduz et al. (2011)
Poyang Lake, China	688 - 825	582 – 691	Xiang and Zhou (2011)
Alalubosa River, Ibadan, Nigeria	307 - 559	140 - 343	Onianwa <i>et al.</i> (2013)
Gege River, Ibadan, Nigeria	1072 - 1240	832 - 943	
Kudeti River, Ibadan, Nigeria	647 – 754	406 - 606	
Ona River, Ibadan, Nigeria	316 - 466	107 - 314	
Ogbere River, Ibadan, Nigeria	249 - 780	230 - 410	
Ogunpa River, Ibadan, Nigeria	355 - 1068	264 – 595	
Onireke River, Ibadan, Nigeria	426 - 647	189 - 603	
Dongping Lake, North China	426 - 730	271 - 513	Chen et al. (2014)
Swart River, Cape Town South Africa	798 – 6877	105 - 2816	present study

Inorganic and Organic Phosphorus in Sediment

The IP component ranged from 13% to 84% (Figure 3) of the TP concentrations in the sediment samples. In 9 of the 14 sampled sediments, IP was the major constituent of TP. The high content of IP for almost all the sampling sites indicated a high rate of mineralisation of OP (Onianwa *et al.*, 2013). Similarly, the HCl-P component of the IP was the predominant form with percentages ranging from 36% to 84%, while the NaOH-P was in the range of 13% to 53%. The concentration of OP ranged from 320

to 3418 ug/g, with an average of 592 ± 270 ug/g, excluding the highest concentration (3418 ug/g). The OP at site 12 (along Swart River) was at least five times that recorded at any other site. Site 12 is confluent, receiving water from the Liesbeek. The Liesbeek River may have contributed to the phosphorus load in the Swart River, as the highest IP was also recorded at this site. Organic-P was the major component of TP in four sites, namely Lange (site 2; 44%), Bridgetown (site 5; 85%), Mowbray golf course (site 10; 61%) and confluence along Swart River towards the ocean (site 13; 51%).

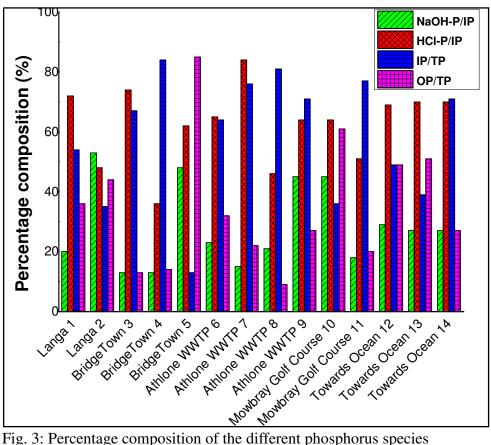


Fig. 3: Percentage composition of the different phosphorus species

Phosphorus Speciation

Inorganic-P was speciated into nonapatite and apatite, those bound to Fe-P and Al-P (NaOH-P) and those bound to Ca-P (HCl-P), respectively. The nonapatite ranged from 112 - 1012 ug/g with a deviation of 322 ± 140 ug/g, while apatite ranged from 574 - 3390 ug/g with an average of 371 ug/g. The speciation results show that HCl-P (Ca-P) was the major component of IP in 12 of the 14 sampling sites. According to studies on the speciation of phosphorus, it has been found that Ca-P and Fe-P can interconvert (Li and Guo, 2006). The extent of the damage done to the river cannot be established without reviewing the sorption capabilities of the sediments from the river. However, if the prevailing conditions persist, in the near future the

sediment beds will be a ready source of biologically available phosphorus under favourable conditions. The mobilisation and re-solubility of phosphorus may lead eutrophication in the aquatic ecosystem; and the resulting anoxic conditions could further increase the resolubilisation of phosphorus (especially inorganic phosphorus) from the sediment bed (DFFE, 2022).

Conclusion

The speciated forms of phosphorus in the sediments collected along the Swart River were evaluated to identify pollution trends and sources. High levels of the different fractions of phosphorus (TP, OP, IP, HCl-P, and NaOH-P) were recorded. The study revealed that the river was polluted from anthropogenic inputs such

as industrial activities, sewage/effluent discharges, and agricultural run-off (use of fertilisers). Speciation results indicate that most of the phosphorus obtained was in the inorganic form, which suggests the mineralisation of OP. The primary composition of IP was bound to Ca-P. The high levels of phosphorus in the sediments of the river can pose a threat if the prevailing conditions are not mitigated. The phosphorus levels may get elevated to severely affect the buffering capacity of the river. Consequently, the water system could experience eutrophication due to high phosphorus levels in the overlying waters.

References

- Acharya, S.S., Panigrahi, M.K., Kurian, J., Gupta, A.K. and Tripathy, S. (2016). Speciation of phosphorus in the continental shelf sediments in the Eastern Arabian Sea. *Continental Shelf Research*, 115: 65-75.
- Aydin, I., Aydin, F., Saydut, A. and Hamamci, C. (2009). A sequential extraction to determine the distribution of phosphorus in the seawater and marine surface sediment. *J Hazard Mater*, 168(2-3):664-669.
- Bains, W., Petkowski, J.J., Sousa-Silva, C. and Seager, S. (2019). Trivalent phosphorus and phosphines as components of biochemistry in anoxic environments. *Astrobiology*, 19(7): 885-902.
- Bastami, K.D., Neyestani, M.R., Raeisi, H., Shafeian, E., Baniamam, M., Shirzadi, A., Esmaeilzadeh, M., Mozaffari, S. and Shahrokhi, B. (2018). Bioavailability and geochemical speciation of

- phosphorus in surface sediments of the Southern Caspian Sea. *Marine Pollution Bulletin*, 126: 51-57.
- Bradford, M.E. and Peters, R.H. (1987). The relationship between chemically analyzed phosphorus fractions and bioavailable phosphorus 1. *Limnology and Oceanography*, 32(5): 1124-1137.
- Chen, Y., Chen, S., Yu, S., Zhang, Z., Yang, L. and Yao, M. (2014). Distribution and speciation of phosphorus in sediments of Dongping Lake, North China. *Environmental Earth Sciences*, 72(8): 3173-3182.
- Coelho, J., Flindt, M., Jensen, H.S., Lillebø, A. and Pardal, M. (2004). Phosphorus speciation and availability in intertidal sediments of a temperate estuary: relation to eutrophication and annual Pfluxes. *Estuarine, Coastal and Shelf Science*, 61(4): 583-590.
- Dan, S.F., Liu, S.-M. and Yang, B. (2020). Geochemical fractionation, potential bioavailability and ecological risk of phosphorus in surface sediments of the Cross River estuary system and adjacent shelf, South East Nigeria (West Africa). *Journal of Marine Systems*, 201: 103244.
- Department of Forestry, Fisheries and the Environment (2022). South African Water Quality Guidelines for Coastal Marine Waters Natural Environment and Mariculture Use. Cape Town, South Africa: The Department of Forestry, Fisheries and the Environment.

- Elser, J.J. (2012). Phosphorus: a limiting nutrient for humanity? *Current Opinion in Biotechnology*, 23(6): 833-838.
- Filippelli, G.M. (1997). Controls on phosphorus concentration and accumulation in oceanic sediments. *Marine Geology*, 139(1-4): 231-240.
- Gunduz, B., Aydın, F., Aydın, I. and Hamamci, C. (2011). Study of phosphorus distribution in coastal surface sediment by sequential extraction procedure (NE Mediterranean Sea, Antalya-Turkey). *Microchemical Journal*, 98(1): 72-76.
- Heathwaite, L. and Sharpley, A. (1999). Evaluating measures to control the impact of agricultural phosphorus on water quality. *Water Science and Technology*, 39(12): 149-155.
- Hou, L., Liu, M., Yang, Y., Ou, D., Lin, X., Chen, H. and Xu, S. (2009). Phosphorus speciation and availability in intertidal sediments of the Yangtze Estuary, China. *Applied Geochemistry*, 24(1): 120-128.
- Jensen, H.S. and Thamdrup, B. (1993). Iron-bound phosphorus in marine sediments as measured bicarbonate-dithionite extraction. Proceedings of the Third International Workshop on Phosphorus Sediments. in Developments in Hydrobiology, 84: 47-59.
- Katsaounos, C.Z., Giokas, D.L., Leonardos, I. D. and Karayannis, M.I. (2007). Speciation of phosphorus fractionation in river sediments by explanatory data

- analysis. *Water Res*, 41(2): 406-418.
- Kovar, J.L. and Pierzynski, G.M. (2009). Methods of phosphorus analysis for soils, sediments, residuals, and waters second edition. *Southern Cooperative Series Bulletin*, 408.
- Li, B. and Guo, B. (2006). Chemical forms of inorganic phosphorus in sediments in the middle of the Yellow River. *Journal of Agro-Environment Science*, 25(6):1607-1610
- Li, C., Yu, H., Tabassum, S., Li, L., Mu, Y., Wu, D., Zhang, Z., Kong, H. and Xu, P. (2018). Effect of calcium silicate hydrates coupled with *Myriophyllum spicatum* on phosphorus release and immobilization in shallow lake sediment. *Chemical Engineering Journal*, 331: 462-470.
- Li, S., Lin, Z., Liu, M., Jiang, F., Chen, J., Yang, X. and Wang, S. (2020). Effect of ferric chloride on phosphorus immobilization and speciation in Dianchi Lake sediments. *Ecotoxicology and Environmental Safety*, 197: 110637.
- Liu, H., Jin, X. and Jing, Y. (1999). Environmental dredging technology of lake sediment. Chinese Engineering Science, 1(1): 81-84.
- Liu, J., Zang, J., Zhao, C., Yu, Z., Xu, B., Li, J. and Ran, X. (2016). Phosphorus speciation, transformation, and preservation in the coastal area of Rushan Bay. Science of The Total Environment, 565: 258-270.
- Liu, S.M., Zhang, J. and Li, D.J. (2004). Phosphorus cycling in sediments

- of the Bohai and Yellow Seas. *Estuarine, Coastal and Shelf Science*, 59(2): 209-218.
- Mainstone, C.P. and Parr, W. (2002). Phosphorus in rivers—ecology and management. *Science of The Total Environment*, 282: 25-47.
- Mao, C., Li, T., Rao, W., Tang, Z., Song, Y. and Wang, S. (2021). Chemical speciation of phosphorus in surface sediments from the Jiangsu Coast, East China: Influences, provenances and bioavailabilities. *Marine Pollution Bulletin*, 163: 111961.
- Mcdonald, G.J., Norton, S.A., Fernandez, I.J., Hoppe, K.M., Dennis, J. and Amirbahman, A. (2019). Chemical controls on dissolved phosphorus mobilization in a calcareous agricultural stream during base flow. *Science of The Total Environment*, 660: 876-885.
- Meng, J., Yao, P., Yu, Z., Bianchi, T.S., Zhao, B., Pan, H. and Li, D. (2014). Speciation, bioavailability and preservation of phosphorus in surface sediments of the Changjiang Estuary and adjacent East China Sea inner shelf. Estuarine, Coastal and Shelf Science, 144: 27-38.
- Murphy, J. and Riley, J. P. (1962). A modified single solution method for the determination of phosphate in natural waters. *Anal Chim Acta*, 27:31-36.
- Onianwa, P., Oputu, O., Oladiran, O. and Olujimi, O. (2013). Distribution and speciation of phosphorus in sediments of rivers in Ibadan, South-Western Nigeria. *Chemical Speciation and Bioavailability*, 25(1): 24-33.

- Oputu, O. and Akharame, M. (2022).

 Assessment of surface water quality within Cape Town, South Africa using NSF water quality index. *International Journal of Energy and Water Resources*:1-13.
- Owa, F. (2014). Water pollution: sources, effects, control and management. *International Letters of Natural Sciences*, 3.
- Pagliosa, P.R., Fonseca, A., Bosquilha, G.E., Braga, E.S. and Barbosa, F.R. (2005). Phosphorus dynamics in water and sediments in urbanized and non-urbanized rivers in Southern Brazil. *Marine Pollution Bulletin*, 50(9): 965-974.
- Paytan, A. and Mclaughlin, K. (2007). Phosphorus in our Waters. *Oceanography*, 20(2): 200-206.
- Penn, M. R. and Auer, M. T. (1997). Seasonal variability in phosphorus speciation and deposition in a calcareous, eutrophic lake. *Marine Geology*, 139(1-4): 47-59.
- Persaud, D., Jaagumagi, R. and Hayton, A. (1993). Guidelines for the protection and management of aquatic sediment quality in Ontario. Ministry of Environment and Energy; Ontario Ministry of the Environment: Toronto, ON, Canada.
- Philander, F.R. (2015). An appraisal of urban agriculture as a livelihood strategy for household food security: a case study of urban food gardens in ward 51, Langa, Cape Town.
- Prüter, J., Leipe, T., Michalik, D., Klysubun, W. and Leinweber, P. (2020). Phosphorus speciation in sediments from the Baltic Sea, evaluated by a multi-method

- approach. *Journal of Soils and Sediments*, 20(3): 1676-1691.
- Rao, J.-L. and Berner, R.A. (1997). Time variations of phosphorus and sources of sediments beneath the Chang Jiang (Yangtze River). *Marine Geology*, 139(1-4):95-108.
- Ruban, V., López-Sánchez, J., Pardo, P., Rauret, G., Muntau, H. and Quevauviller, P. (1999). Selection and evaluation of sequential extraction procedures for the determination of phosphorus forms in lake sediment. *Journal of Environmental Monitoring*, 1(1): 51-56.
- Ruban, V., López-Sánchez, J., Pardo, P., Rauret, G., Muntau, H. and Ouevauviller, P. (2001).Harmonized protocol and certified material reference for the determination of extractable phosphorus contents of freshwater sediments-a synthesis of recent works. Fresenius' Journal of Analytical Chemistry, 370(2-3): 224-228.
- Ruttenberg, K.C. (1992). Development of a sequential extraction method for different forms of phosphorus in marine sediments. *Limnology and Oceanography*, 37(7): 1460-1482.
- Slomp, C. P., Malschaert, J. and Van Raaphorst, W. (1998). The role of adsorption in sediment-water exchange of phosphate in North Sea continental margin sediments. *Limnology and Oceanography*, 43(5): 832-846.
- Spivakov, B.Y., Maryutina, T. and Muntau, H. (1999). Phosphorus speciation in water and sediments. *Pure and Applied Chemistry*, 71(11): 2161-2176.

- Sundby, B. R., Gobeil, C., Silverberg, N. and Alfonso, M. (1992). The phosphorus cycle in coastal marine sediments. *Limnology and Oceanography*, 37(6): 1129-1145.
- Thin, M.M., Sacchi, E., Setti, M. and Re, V. (2020). A Dual Source of Phosphorus to Lake Sediments Indicated by Distribution, Content, and Speciation: Inle Lake (Southern Shan State, Myanmar). Water, 12(7): 1993.
- Xiang, S.-L. and Zhou, W.-B. (2011). Phosphorus forms and distribution in the sediments of Poyang Lake, China. *International Journal of Sediment Research*, 26(2): 230-238.
- Yamada, H. and Kayama, M. (1987).

 Distribution and dissolution of several forms of phosphorus in coastal marine-sediments.

 Oceanologica Acta, 10(3): 311-321.
- Yang, B., Lan, R.-Z., Lu, D.-L., Dan, S.F., Kang, Z.-J., Jiang, Q.-C., Lan, W.-L. and Zhong, Q.-P. (2019). Phosphorus biogeochemical cycling in intertidal surface sediments from the Maowei Sea in the northern Beibu Gulf. *Regional Studies in Marine Science*, 28: 100624.
- Yue, L. and Ji-Jin, Y. (1999). Geochemical characteristics of phosphorus near the Huanghe River estuary. *Chinese Journal of Oceanology and Limnology*, 17(4): 359-365.
- Zhou, F., Gao, X., Yuan, H., Song, J., Chen, C.-T. A., Lui, H.-K. and Zhang, Y. (2016). Geochemical forms and seasonal variations of phosphorus in surface sediments of the East China Sea shelf. *Journal of Marine Systems*, 159: 41-54.